

# **SILICON NANOPARTICLES COATED WITH DNA-DERIVED CARBON AS HIGH-PERFORMANCE ANODE MATERIAL FOR LITHIUM ION BATTERIES**

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Recently, silicon has been attracting great interest as an anode material for lithium ion-batteries because of its excellent ability to store lithium ions ( $\text{Li}^+$ ) when compared to the conventionally used carbon materials. However, silicon-based electrode material experiences a drastic volume change (*i.e.*, expansion and contraction) during charge/discharge cycles, leading to a catastrophic failure of the anode. The purpose of this study is to sustain the silicon's high ability to absorb or release  $\text{Li}^+$  through the homogeneous coating of DNA-derived amorphous carbon. Silicon nanoparticles coated with double-stranded DNA are thermally treated at  $800^\circ\text{C}$  in an argon atmosphere in order to convert the DNA into amorphous carbon. We have observed the homogeneously coated thin, disordered carbons on silicon nanoparticles. Finally, we have verified the effective buffer-like function of DNA-derived carbon by the long-term stable electrochemical behavior of silicon-based anodes in lithium ion batteries.

# Silicon nanoparticles coated with DNA-derived carbon as high-performance anode material for lithium-ion batteries

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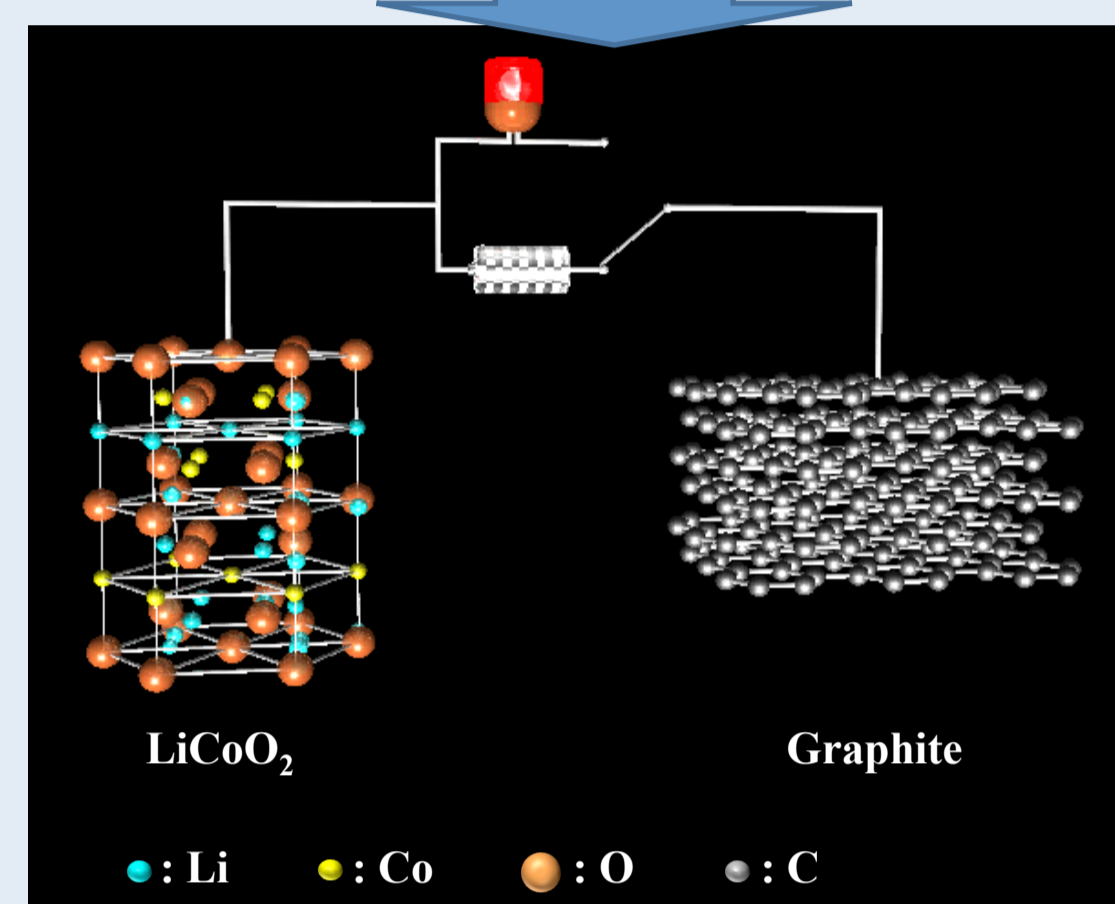
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2. Endo-Hayashi-Kim Lab, Shinshu University, Engineering Division

## Background



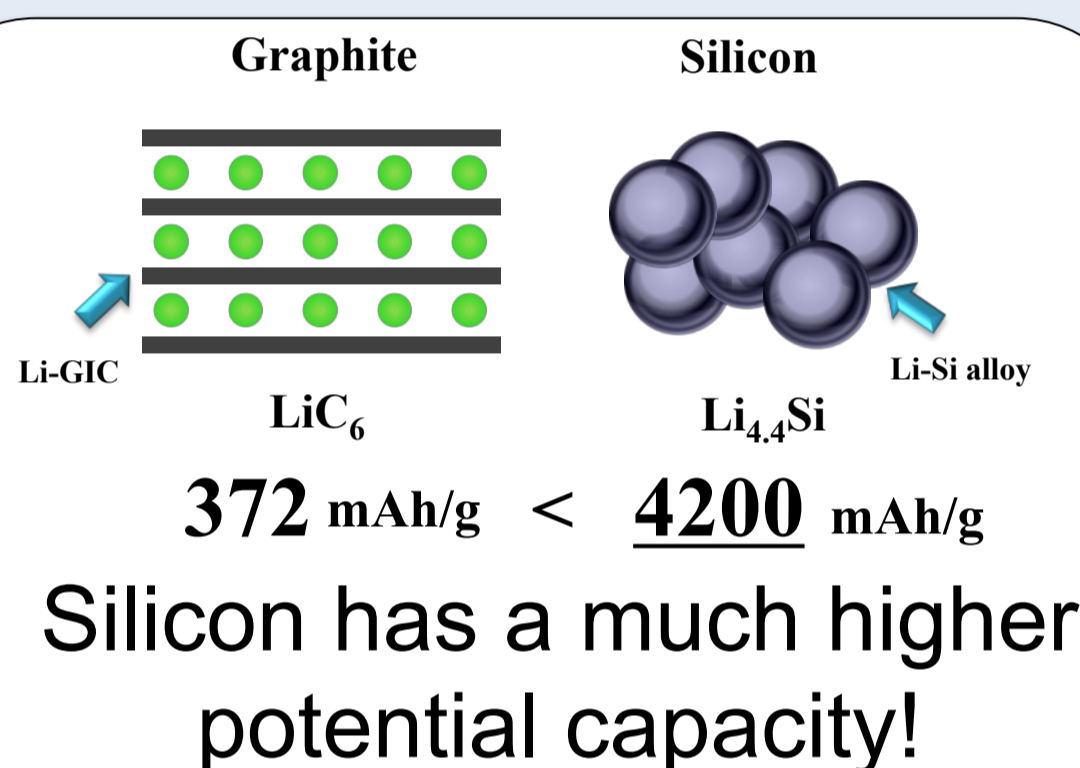
### Lithium-Ion Batteries

- The dominant power sources for portable electric devices and EVs.
- Graphite generally used as the anode material.
- However, higher-capacity materials are now being investigated.



### Higher Capacity

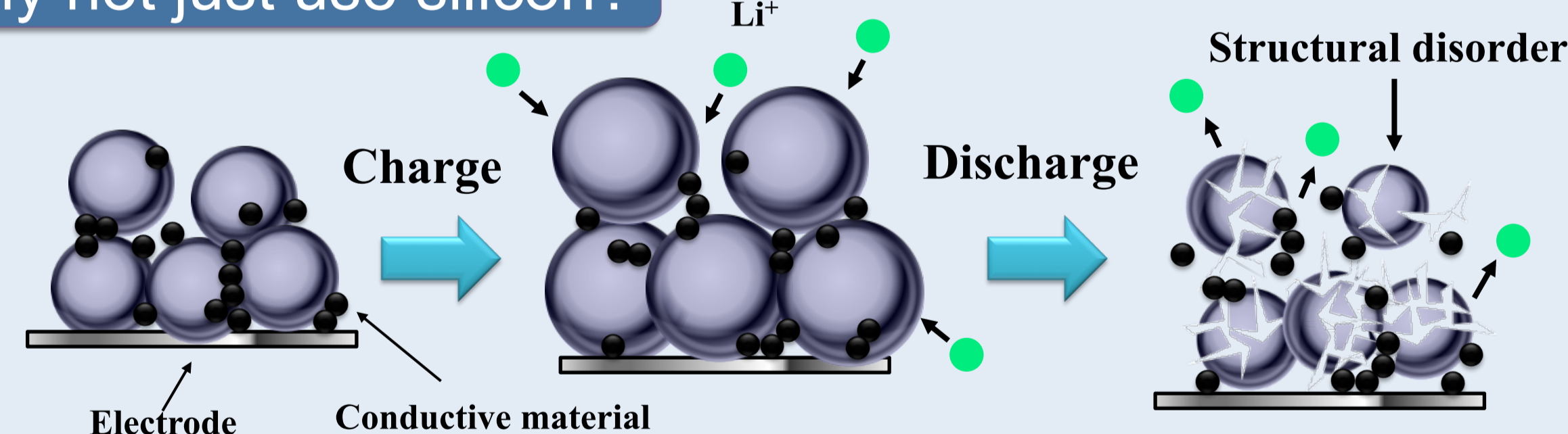
- More time b/w charges
- Longer-lasting laptops
- Longer-running EVs



## Purpose

**Objective :** Increase capacity of a Lithium-ion battery by using carbon-coated silicon instead of graphite as an anode material.

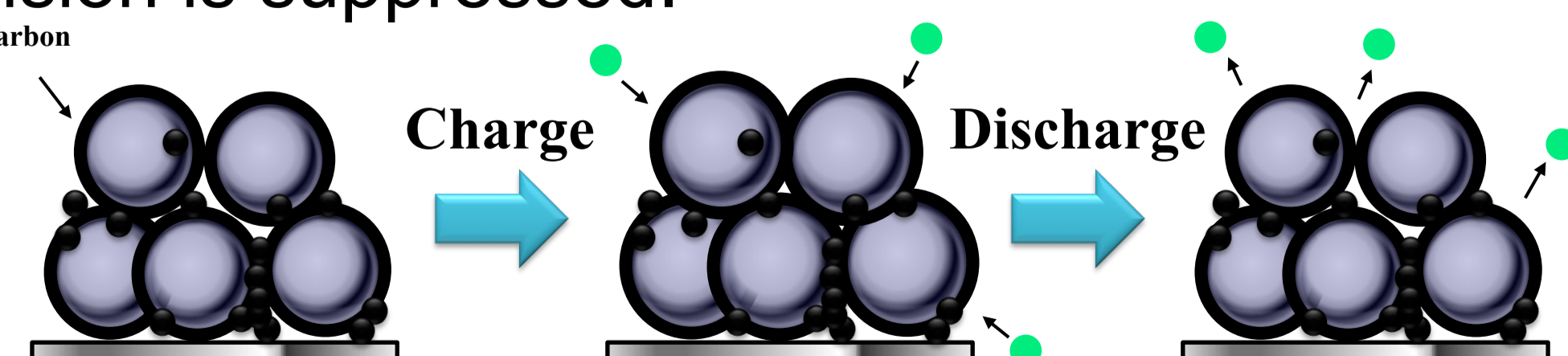
### Why not just use silicon?



Silicon anode expands drastically as it absorbs lithium.

Loss of conductivity; Capacity decreases.

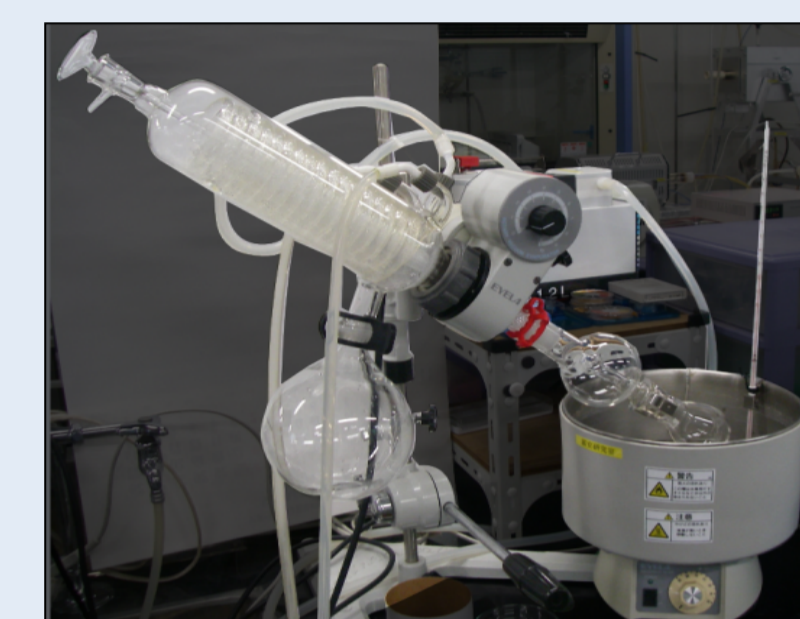
**Solution:** Silicon is coated with DNA-derived carbon and expansion is suppressed.



## Method

### Material creation

- Si nanoparticles mixed with fish DNA in water (Si:DNA = 2:1).
- Si-DNA mixture extracted by evaporator.
- DNA converted into amorphous carbon by heat treatment at 800°C.



Evaporator



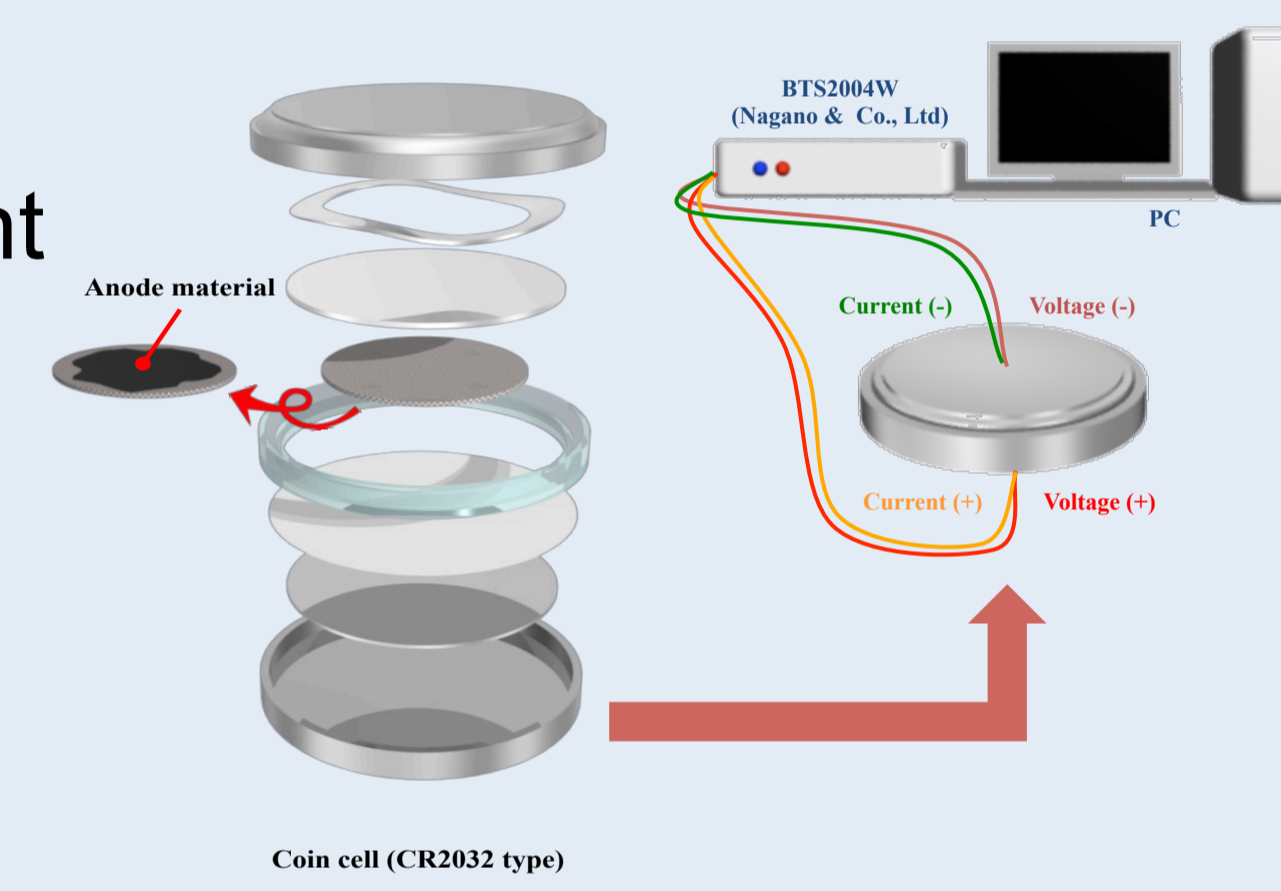
Electric furnace

### Analysis

- Raman
- TEM
- Battery Measurement



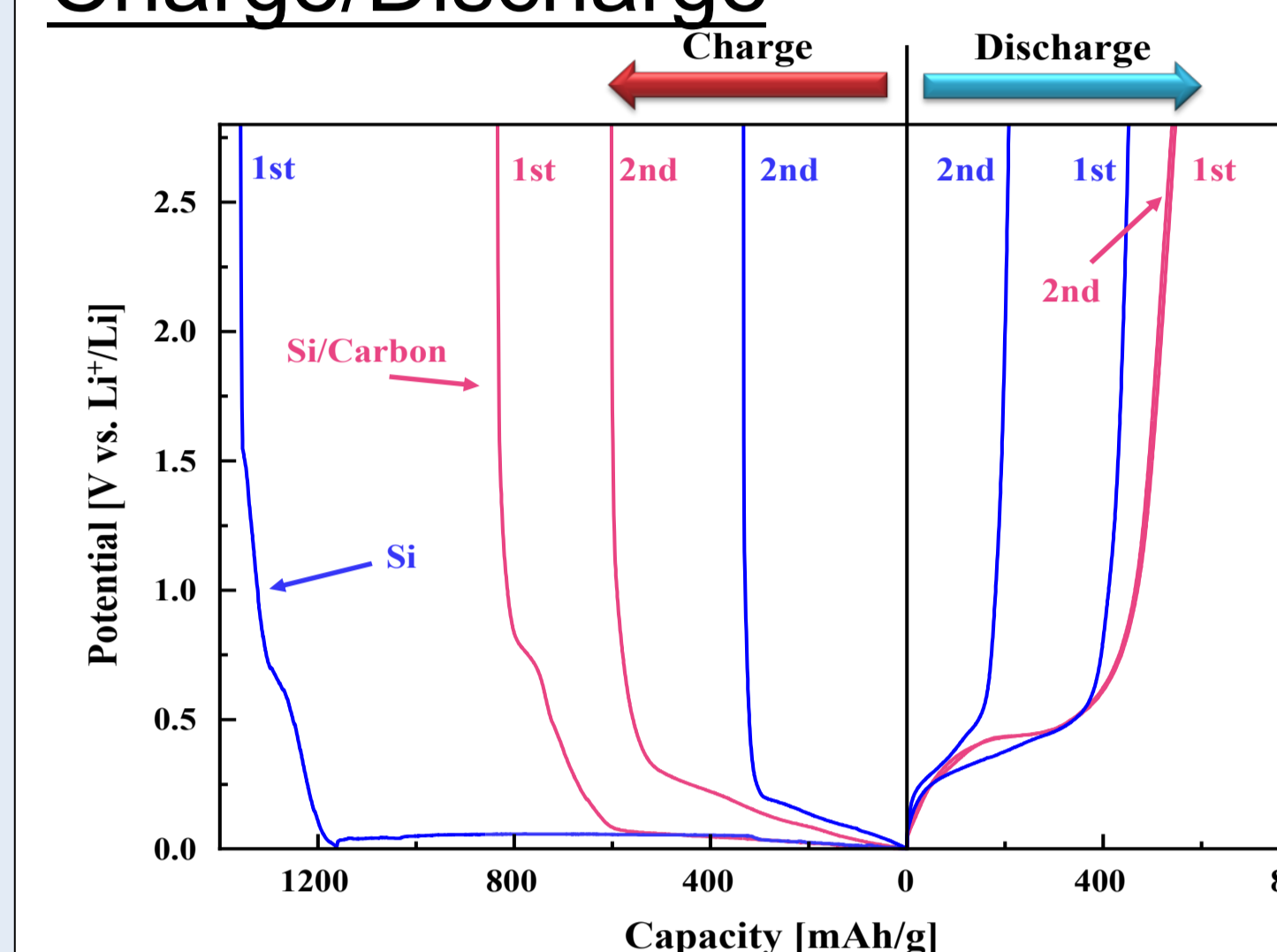
TEM



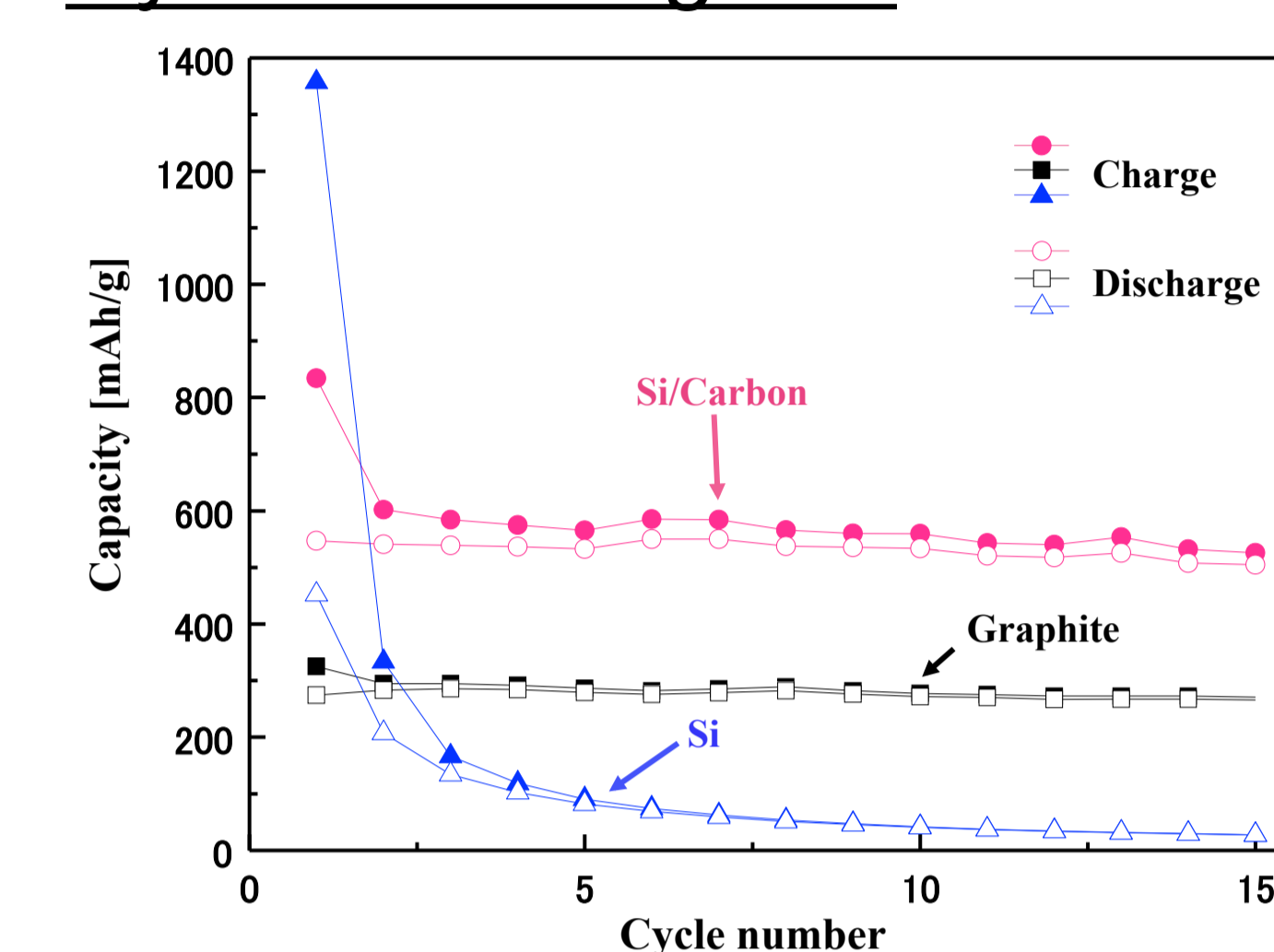
Battery measurement

## Results - Batteries

### Charge/Discharge



### Cycles 1 through 15



### Si-Carbon anodes

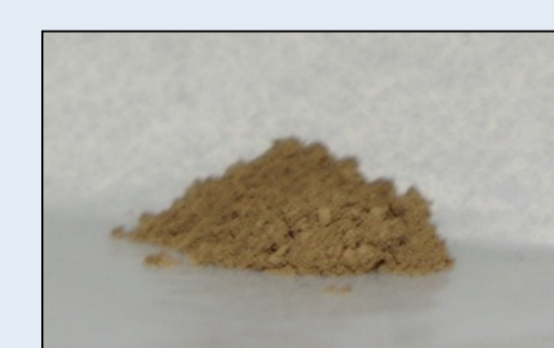
#### Capacity

- Exhibit the highest discharge capacity.
- Irreversible capacity lower than that of Si batteries.

#### Cycle Life

- An overall capacity of around 600 mAh/g.
- About the same cycle life as graphite anodes.

## Results - Materials

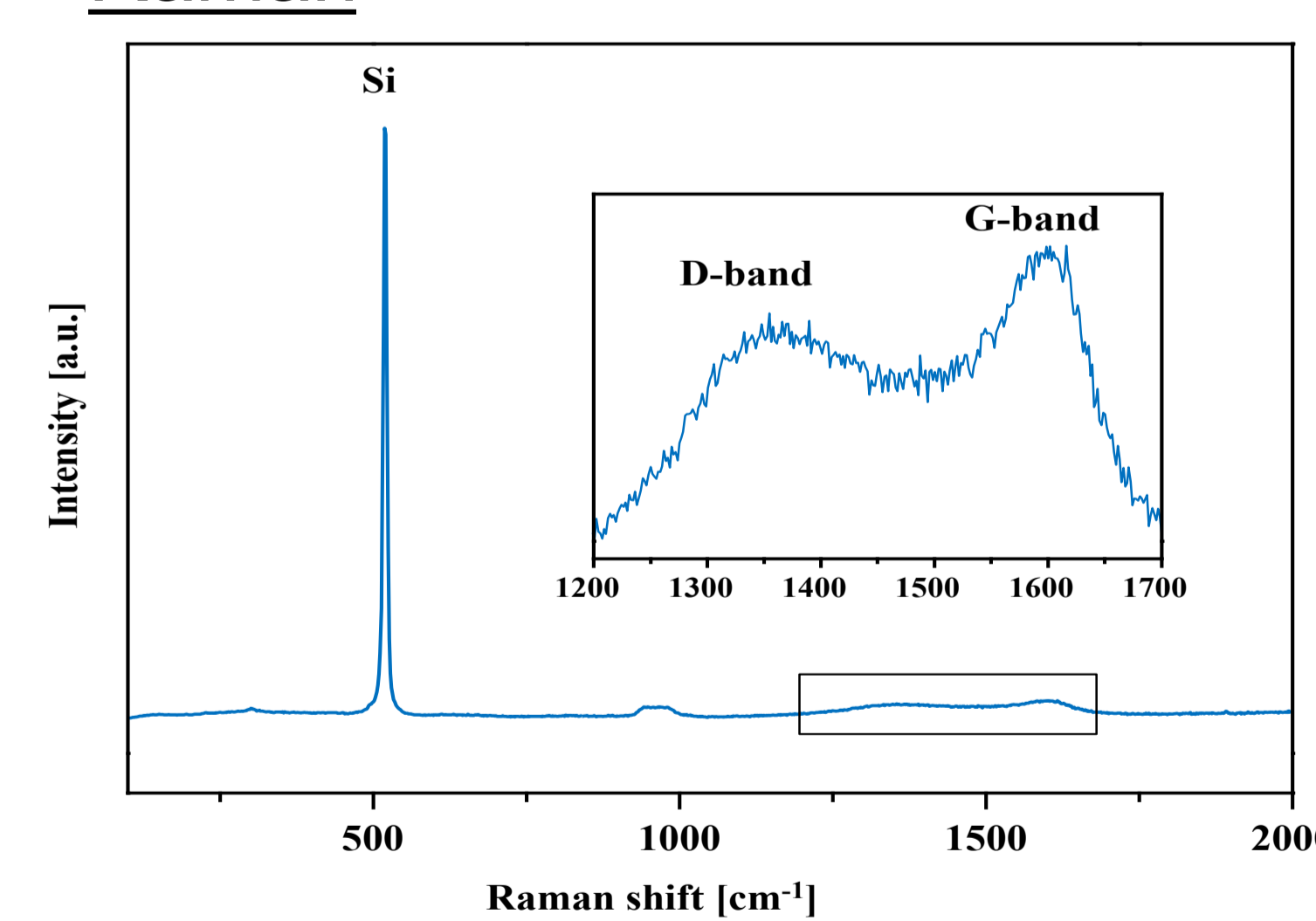


Before



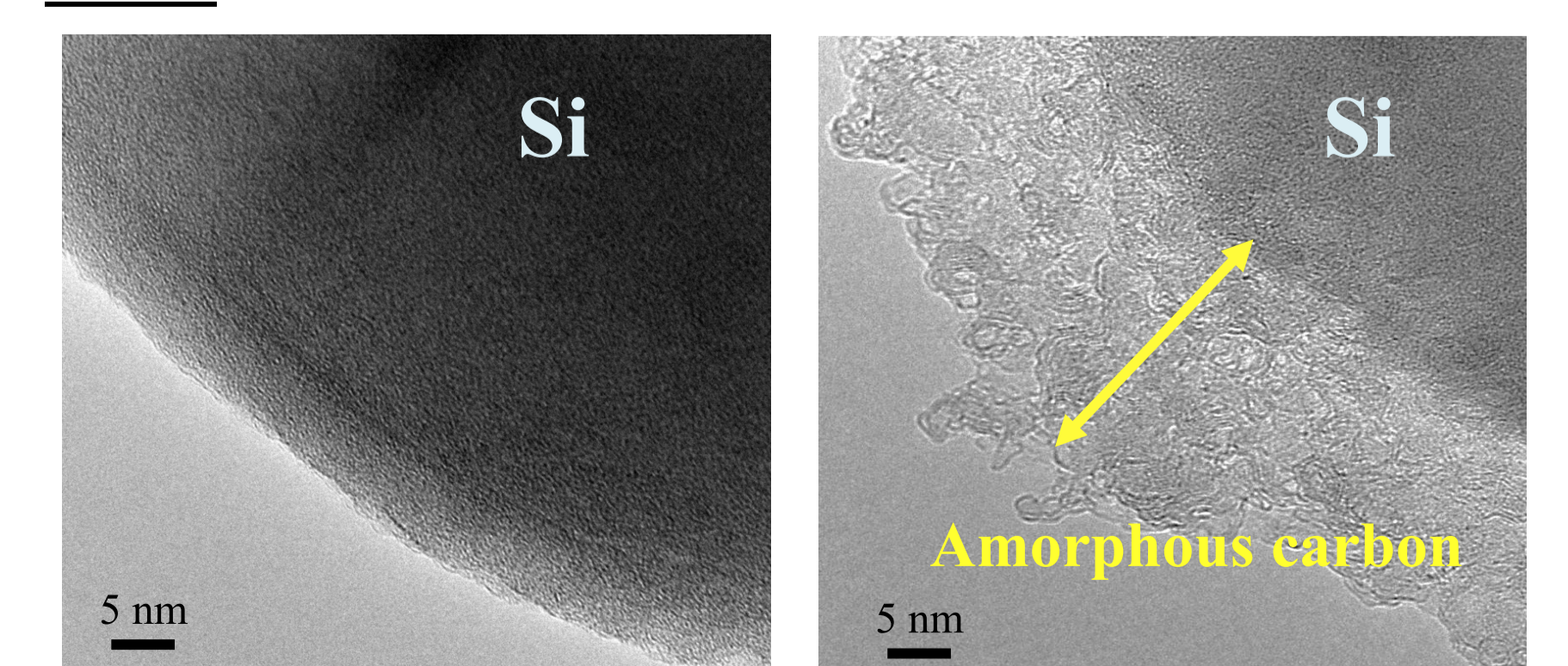
After

### Raman



- Both silicon and carbon were detected.

### TEM



- Silicon was firmly coated with amorphous carbon.

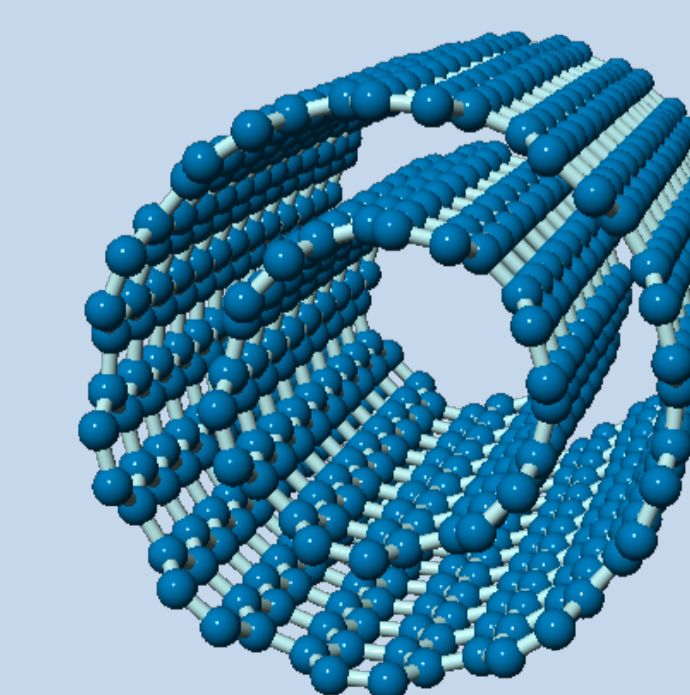
## Conclusion

Expansion and contraction during charge/discharge suppressed by carbon coating. The resulting Si-Carbon anodes:

- Are more stable-cycling than silicon-based batteries.
- Have twice the capacity of graphite-based batteries.

### Future Projects

- Alter temperature of heat treatment: 600 or 1000 °C
- Add double-walled carbon nanotubes to improve anode conductivity and structural integrity.
- Reduce the size of the silicon nanoparticles.



## Acknowledgements

Research conducted at Shinshu University's Endo-Hayashi-Kim Laboratory as a participant of the 2012 NanoJapan program, sponsored by NSF-PIRE and Rice University.